

Exposure to Radon

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Radon

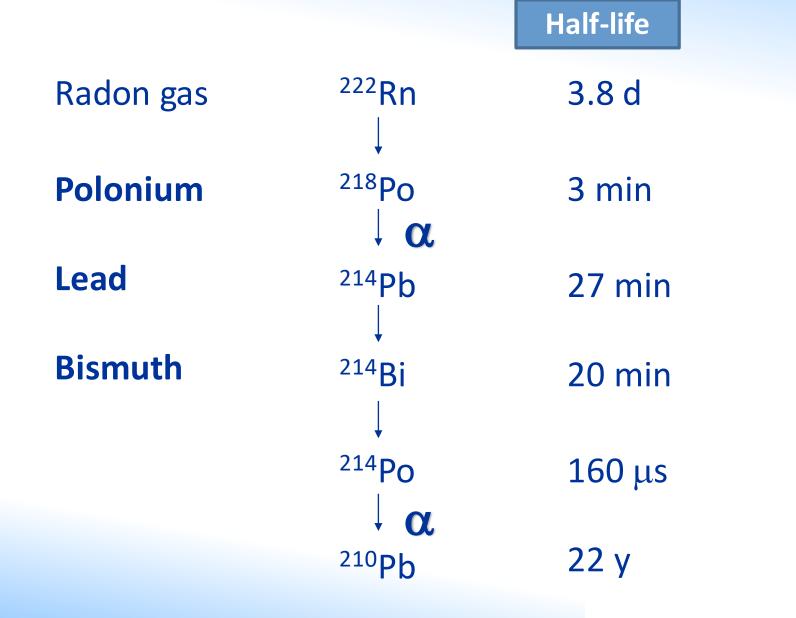


Radon (²²²Rn)

- inert gas
- naturally radioactive
- Part of ²³⁸U decay series
- ²²⁶Ra as its immediate parent
- Half-life 3.8 days
- Short-lived decay progeny: ²¹⁸Po, ²¹⁴Pb, ²¹⁴Bi, ²¹⁴Po
- Main exposure pathway Escape of ²²²Rn to the air and subsequent inhalation
- Secondary exposure pathway Dissolution of ²²²Rn in groundwater and subsequent ingestion (or inhalation on release from water into the air)

Decay chain of Rn-222





Thoron



²²⁰Rn (thoron)

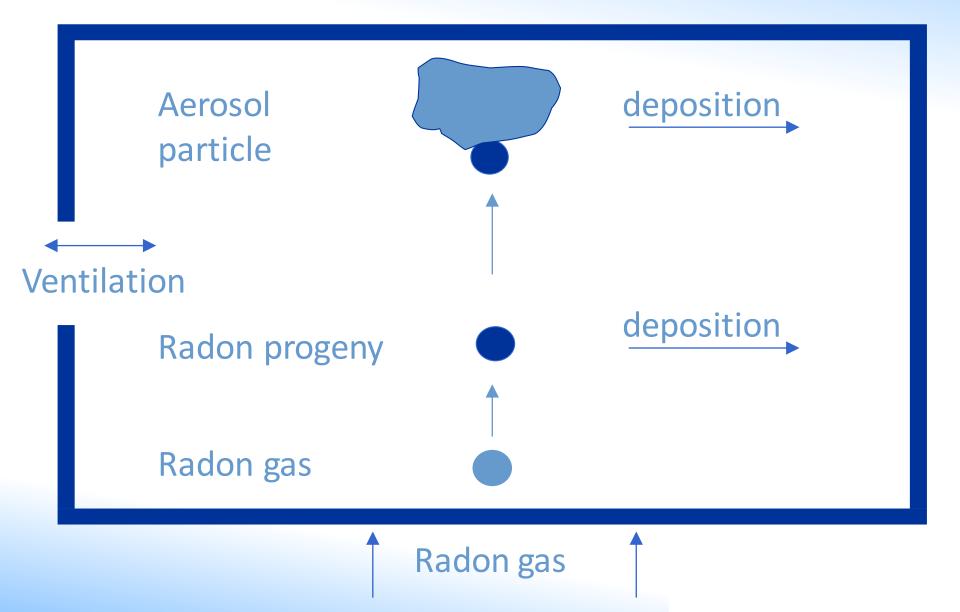
- Part of ²³²Th decay series
- ²²⁴Ra as its immediate parent
- Gaseous form
- Short half-life ~56 seconds
- Rapidly decays into thoron progeny
- Progeny have a half-life of up to 10.66 hours
- Exposures from progenies ²¹²Pb and ²¹²Bi
- Inhalation exposure is the main concern particularly in thorium processing facilities
- Only transported over short distances
- In the absence of thoron gas the progeny concentrations decline slowly



- Contribution of ²²²Rn itself to the lung dose is very small
 - Most of the inhaled gas is breathed out again
- Short-lived progeny are responsible for about 99% of the dose:
 - Atoms attach to condensation nuclei and dust particles in the air that is inhaled
 - Unattached progeny also inhaled
 - Deposited along the various airways in the bronchial tree
 - Most exposure comes from alpha particles
 - Contributions from beta and gamma emissions are small in comparison

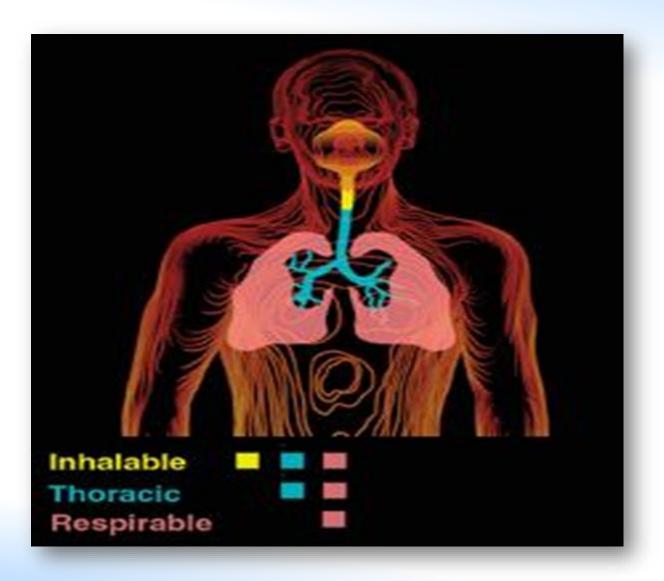
Formation of radon progeny aerosol





Respiratory tract







- Workplaces where exposure to other U, Th series radionuclides already needs to be controlled as a planned exposure situation:
 - Exposure to ²²²Rn is subject to the requirements for planned exposure situations, regardless of the radon concentration
 - Regulatory control must be considered
 - Any necessary control measures for ²²²Rn must be included in the RPP

Application of the Standards to ²²²Rn in workplaces



- Workplaces where exposure to other U, Th series radionuclides does not need to be controlled:
 - Above-ground workplaces:
 - High ²²²Rn levels can usually be reduced to well below the reference level by remedial action:
 - Regulatory control as a planned exposure situation is therefore unlikely to be needed
 - Underground workplaces
 - ²²²Rn may be more difficult to control by remedial action:
 - High ²²²Rn ingress via rock surfaces or water, limitations on ventilation
 - Regulatory control as a planned exposure situation is necessary if the ²²²Rn concentration remains above the reference level



- Short-lived ²²²Rn progeny are unlikely to be in equilibrium with the parent
- Therefore, for radiation protection purposes, special quantities are needed for ²²²Rn progeny:
 - Concentration in air
 - The resulting exposure
- These special quantities are derived from a quantity known as the 'potential alpha energy'



The potential alpha energy ε_p of a <u>single atom</u> of a short-lived ²²²Rn progeny radionuclide is the total alpha energy emitted by that atom during its progressive radioactive decay down to, but not including, the relatively long-lived radionuclide ²¹⁰Pb



Potential alpha energy concentration (PAEC) in air:

For any mixture of short-lived ²²²Rn progeny in air, the contribution of each radionuclide to the PAEC is its potential alpha energy per unit activity (ε_p/λ) given in the previous.

Table multiplied by its activity concentration, *c*. The total PAEC is then the sum of these individual contributions

$$PAEC = \sum_{i} c_{i} \left(\varepsilon_{p,i} / \lambda_{i} \right)$$



- In practice, the progeny will rarely be in equilibrium
- The PAEC will therefore be some fraction of 5.56×10^{-9} J/m³.
- This fraction is called the <u>equilibrium factor</u> (F)
- For a ²²²Rn concentration of 1 Bq/m³, the PAEC of any given non-equilibrium mixture will be:

$$PAEC\left(J/m^3\right) = 5.56 \times 10^{-9} \times F$$



The equilibrium factor can vary from:

-Zero (radon gas diffusing out of the grain surface of the mineral)

to

-1 (radon gas after three hours in stagnant air)

An equilibrium factor of 0.4 is usually taken as a default value

The older the air the higher the progeny concentrations



F is a measure of the degree of dis-equilibrium between radon gas and its progeny

<i>F</i> =1		<i>F</i> =0.4	
Nuclide	Bq m ⁻³	Nuclide	Bq m⁻³
²²² Rn gas	1.0	²²² Rn gas	1.0
²¹⁸ Po	1.0	²¹⁸ Po	0.7
²¹⁴ Pb	1.0	²¹⁴ Pb	0.4
²¹⁴ Bi	1.0	²¹⁴ Bi	0.3

The value of *F* depends on the ventilation rate :

Indoors : $F \approx 0.4$ Natural ventilation Mines : $F \approx 0.2$ Forced ventilation



- Multiply PAEC (J/m³) by occupancy (h) to give exposure in units of J·h·m⁻³
- The PAEC will vary with time, so the exposure has to be calculated as an integral over time (which in practice is the average concentration):
- Exposure usually determined over 1 year
- A default annual occupancy time of 2000 h may be assumed for workplaces (often conservative)

Equilibrium equivalent concentration



- Equilibrium equivalent concentration (EEC) is an alternative to using PAEC
- EEC is the concentration of ²²²Rn in equilibrium with its progeny that would give the same PAEC as the actual nonequilibrium mixture
- The EEC is related to the PAEC by a constant factor:

$$PAEC$$
 (in J/m³) = 5.56×10⁻⁹ × EEC (in Bq/m³)

Historical unit of exposure (for awareness)



 Exposure is sometimes expressed in historical units of 'working level month' (WLM)

 $1 \text{ WLM} = 0.00354 \text{ J} \cdot \text{h} \cdot \text{m}^{-3}$

- Commonly used in uranium mines
- The use of this unit is discouraged

²²²Rn gas concentration as a surrogate for ²²²Rn progeny concentration



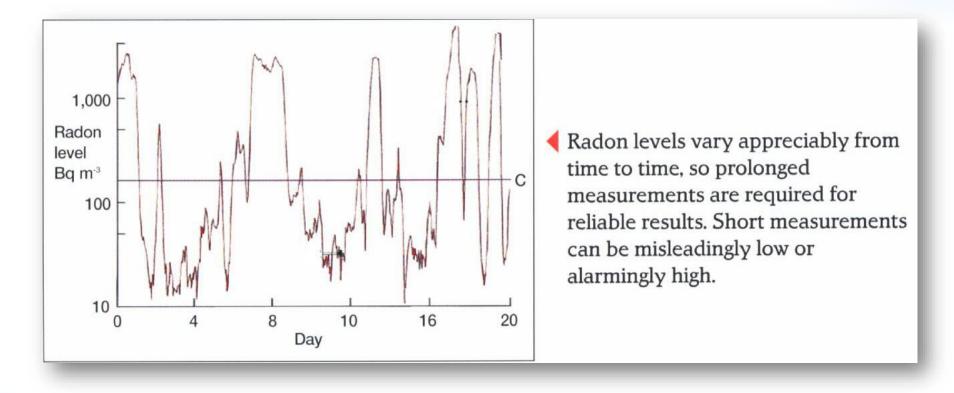
- Measurement of ²²²Rn gas concentration is simpler e.g. measurements in buildings over extended periods
- Passive radon track-etch devices are small, simple, robust, inexpensive
- An equilibrium factor has to be assumed
- An assumed value of 0.4 is usually adequate for buildings, but can be significantly different in underground mines
- Usual approach in buildings
- It is also acceptable in many underground workplaces

Measurement



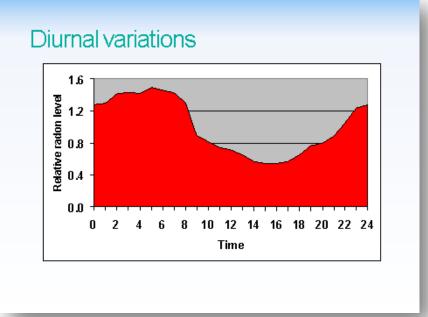
- A wide variety of measurement methods are available
- Either the gas or the progeny can be measured
- Rapid temporal variations can occur
- Integrated (long term measurements) are preferred due to the variability in the gas and progeny concentrations
- Equipment calibration is important





Variations





Seasonal variations

A M

J F M

Ĵ

Month

А

J

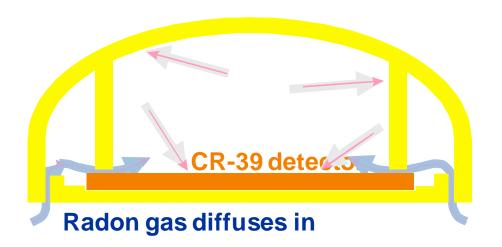
S 0

N D

Radon gas passive monitor



• Some alpha particles emitted by radon and daughters strike detector





Workplace monitoring – radon gas



- Grab sampling: Lucas cells
- Grab sampling: pulse ionization chambers
- Grab sampling: two filter method
- Integrated measurements using nuclear track detectors (radon cup)
- Continuous monitoring

Types of radon detectors





Workplace monitoring – radon progeny



- Air is drawn through a filter
- The particulate progeny are collected
- The gross alpha (and/or beta) decay is measured

or

Alpha spectrometry can be used to measure each radon progeny





- Sampling strategies require to be statistically based random/ extensive screening
- Usually assessed by grab sampling (10–30 minutes)
- Requires a documented monitoring programme
- The monitoring frequency depends upon:
 - The level of concentration
 - The variability of concentration



The sampling frequency is increased when:

- measured concentrations exceed the usual range
- major changes are made to the ventilation system
- reference levels are exceeded
- the effectiveness of corrective actions is to be assessed
- ingress of radon into the working area is suspected

Individual passive radon monitoring

Radon gas

- Personal radon monitors are usually passive devices
- Passive monitors rely on radon diffusion into the detector
- Track-etch detectors are solid state devices, i.e. plastic
- Radon decays through alpha decay
- The alpha particle marks the detector
- Passive detectors are integrating devices
- Their exposure periods can range up to three months







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Individual active radon progeny monitoring

Radon progeny

- Active monitors use pumps and filters
- Known volumes of air are sampled
- Short term integrating devices, e.g. full shift
- Collects the particulate radon progeny

Types of detector:

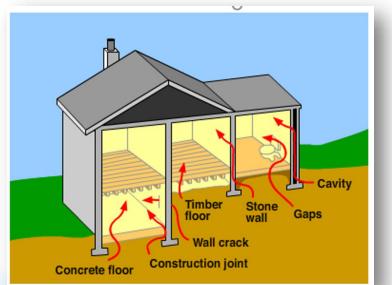
- Solid state silicon detectors
- Track etch





Radon in buildings

- Ingress mainly from underlying soil through cracks in floor:
 - Indoor air pressure usually lower than outside
 - Indoor air is warmer convective flow
 - Wind blowing over chimneys and other openings
 - Other factors affecting ingress through floor:
 - Relative humidity
 - Soil moisture
 - etc.





Radon in buildings (2)

- Other (usually less prominent) forms of ingress:
 - Emission from building materials:
 - Type of material
 - Porosity
 - ²²⁶Ra concentrations
 - Emission from water supply:
 - ²²²Rn concentration in water usually low
 - May be higher if originating from groundwater
 - Minerals and raw materials in workplaces, depending on:
 - ²²⁶Ra concentrations, surface area (particle size), porosity, radon emanation coefficient

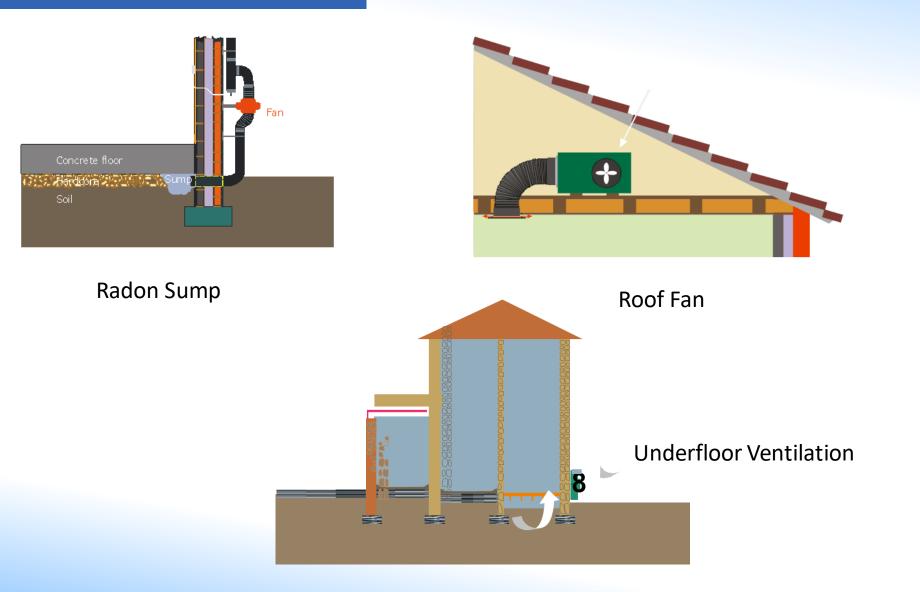


Radon in buildings (3)

- Indoor concentrations:
 - Variation between countries:
 - Geology
 - Climate
 - Construction techniques
 - Domestic habits
 - National mean values: 7–200 Bq/m³
 - Population weighted worldwide mean: 39 Bq/m³
 - Variation within individual countries
 - High background areas: 112 2745 Bq/m³
 - Up to 84 000 Bq/m³ in some parts of northern Europe



Mitigation - buildings





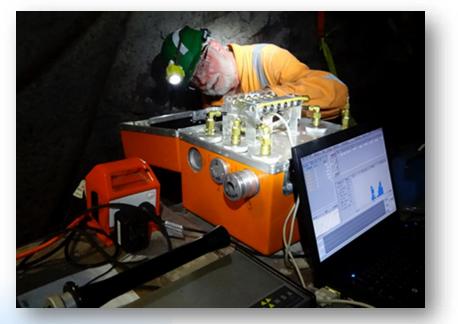
Radon in underground workplaces

- Examples:
 - Tunnels
 - Basement storage & parking facilities
 - Underground water treatment plants
 - Caves and former mines open to the public
 - Spas
 - Underground mines
- Causes of high concentrations:
 - Large areas of interface with the ground
 - Practical limitations on ventilation
 - Ingress of water laden with ²²²Rn
 - High ²²⁶Ra concentrations in the ground (certain mines)



Radon in underground workplaces

- Concentrations:
 - Caves and mines open to the public: 20 to >20 000
 Bq/m³
 - Workplaces in tunnels: 200 to 7000 Bq/m³
 - Even higher values possible in underground mines
 - Particularly uranium mines





Calculation of dose

- Reference levels are expressed in terms of ²²²Rn concentrations rather than dose
- Conversion to dose is however necessary when exposure to ²²²Rn progeny is treated as a planned exposure situation
 - UNSCEAR recommend a dose factor of 9 nSv per Bq·h·m⁻³
 - 1.6 Sv per J·h·m⁻³ potential alpha energy exposure
- Dose due to ²²²Rn progeny has to be added to doses from external gamma and inhaled or ingested dust

Thoron



- Thoron (²²⁰Rn) is not normally of concern in NORM industries, except where material with a high thorium content is processed, e.g. monazite
- Short lived progeny are likely to be severely out of equilibrium with the parent
- In enclosed workplaces, the short half-life of thoron (55.6 s) means that the spatial distribution of thoron is much different from that of its progeny
 - Assessment of equilibrium factor is difficult
 - For dose assessment, an approach based on measurement of progeny concentration is easier and more appropriate than an approach based on measurement of thoron concentration

Thoron



- Dose factors:
 - The UNSCEAR value for dose from inhalation of thoron is 40 nSv per Bq·h·m⁻³ equilibrium equivalent exposure
 - Using this value:
 - A thoron progeny potential alpha energy exposure of 1 mJ·h·m⁻³ gives a committed effective dose of about 0.5 mSv

(Compared with 1.6 mSv for ²²²Rn progeny)





- Radon is everywhere and concentrations can vary significantly
- Exposure is subject to regulation
- Dose comes from the radon progeny
- Special quantities for concentrations and exposures
- Many measurement methods exist and vary from simple to complex
- Selection depends upon the situation and the possible concentration